BORATES

Borate minerals are minerals that contain a borate anion group. Simple borate anion is BO_3 and has a -3 charge. This means each borate anion requires a +3 cation to generate a non-charged stable compound or mineral. But there are also more complex boron-oxygen combinations like B_2O_5 amd B_3O_6 .

59. Ulexite - NaCaB₅O₆(OH)₆·5H₂O

Color Crystal Habit Crystal System Cleavage Hardness Luster/Streak
Use (if any) Other





Ulexite is also known as <u>TV</u> rock due to its unusual optical characteristics. The fibers of ulexite act as <u>optical</u> <u>fibers</u>, transmitting light along their lengths by internal reflection. When a piece of ulexite is cut with flat polished faces perpendicular to the orientation of the fibers, a good-quality specimen will display an image of whatever surface is adjacent to its other side



CARBONATES

Carbonate minerals are minerals that contain a carbonate anion group. Simple borate anion is ${\rm CO_3}$ and has a -2 charge. This means each borate anion requires a +2 cation to generate a non-charged stable compound or mineral. There are many very common carbonate minerals. You are responsible for knowing 5 of them, two of which, aragonite and calcite, have the same chemical composition

5.	Aragonite	- CaCO ₃
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Color Crystal Habit Crystal System Cleavage Hardness Luster/Streak
Use (if any) Other





Near surface, low temperature phase

5. Calcite - CaCO₃

Color Crystal Habit Crystal System Cleavage Hardness Luster/Streak
Use (if any) Other







Dogtooth crystals

CARBONATES - continued

19.	Dolomite	CaMg($(CO_3)_2$
		O.	3,2

/





7. Azurite $Cu_3(CO_3)_2(OH)_2$

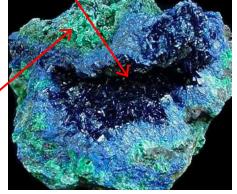
Color -	
Crystal Habit -	
Crystal System -	
Cleavage -	
Hardness -	
Luster/Streak	/
Use (if any) -	
Other	





36. Malachite Cu₂CO₃(OH)₂

Color Crystal Habit Crystal System Cleavage Hardness Luster/Streak
Use (if any) Other







HALIDES

The **halide mineral** class include those **minerals** with a dominant **halide** anion (F⁻, Cl⁻, Br⁻ and I⁻). Each carries a -1 charge that must be balanced. You are only responsible for two.

22. Fluorite – Ca	F ₂		P.	
Color - Crystal Habit - Crystal System - Cleavage -				
Hardness - Luster/Streak Use (if any) - Other	/	<i>y</i>		

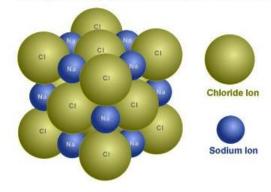
30. Halite - NaCl

Color Crystal Habit Crystal System Cleavage Hardness Luster/Streak
Use (if any) Other

Evaporite mineral



Arrangement of Sodium and Chloride Ions in Halite



HYDROXIDES

A group of minerals combining metallic elements with either water (H₂O) or a hydroxyl goup (OH) or both.

8. Bauxite – Al(OH)₃ or AlO(OH)

Gibbsite, Diaspore

Color -	
Crystal Habit -	
Crystal System -	
Cleavage -	
Hardness -	
Luster/Streak	/
Use (if any) -	
Other	



Almost all aluminum produced has come from bauxite, which is actually a combination of minerals.



OXIDES

An **oxide** is a chemical compound that contains at least one oxygen atom and one other element in its chemical formula.

17. Corundum - Al₂O₃

Color Crystal Habit Crystal System Cleavage Hardness Luster/Streak
Use (if any) Other





Rubies and sapphires are gem corundum.

OXIDES — continued — Iron OXIDES

24. Goethite/L	imonite (FeO	(OH)	
Color - Crystal Habit - Crystal System - Cleavage - Hardness - Luster/Streak Use (if any) - Other			
31. Hematite	Fe ₂ O ₃		
Color - Crystal Habit - Crystal System - Cleavage - Hardness - Luster/Streak Use (if any) - Other			
35. Magnetite Color - Crystal Habit - Crystal System -	Fe ₃ O ₄	-	
Cleavage - Hardness - Luster/Streak Use (if any) - Other	/		

PHOSPHATES

Phosphate minerals are those minerals that contain the tetrahedrally coordinated phosphate (PO_4^{3-}) anion along with the freely substituting **arsenate** (AsO_4^{3-}) and **vanadate** (VO_4^{3-}). **Chlorine** (Cl^-), fluorine (F^-), and **hydroxide** (OH^-) anions also fit into the crystal structure.



4. Apatite $Ca_5(PO_4)_3(F,CI,OH)$

Color - ______ Crystal Habit - _____ Crystal System - ____ Cleavage - ____ 5 on Mohs scale Luster/Streak _____ /___ Use (if any) - ____ Other





